

IN THE ABSTRACT:

Please rewrite the Abstract as follows:

Carbon dioxide containing gas is blown into a slurry containing 100 to 400 g/L of calcium hydroxide ~~slurry having a calcium hydroxide concentration of 100 to 400 g/L~~ obtained by wet slaking calcined lime with 150 to 400 ml of 4 N hydrochloric acid activity (value at 3 minutes) for reaction ~~of 150 to 400 mL to allow them to react until~~ the carbonation rate becomes 50 to 85%, then 1 to 20% by volume of the calcium hydroxide slurry is added, and carbon dioxide containing gas is introduced further blown in to terminate the reaction. The product ~~calcium carbonate obtained by this method~~ is precipitated calcium carbonate aggregates having a secondary particle diameter of 1 to 10 μm and consisting of primary particles having a long diameter of 0.5 to 3.0 μm , a short diameter of 0.1 to 1.0 μm and an aspect ratio of less than 3, has ~~superior characteristics due to a~~ BET specific surface area in the range of 8 to 20 m^2/g and a pore volume in the range of 1.5 to 3.5 cm^3/g , ~~can be uniformly dispersed in pulp fiber, and thereby can be used to manufacture high bulk paper.~~